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Communication: Re-entrant limits of stability of the liquid phase and the Speedy scenario in colloidal model systems

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A re-entrant gas-liquid spinodal was proposed as a possible explanation of the apparent divergence of the compressibility and specific heat off supercooling water. Such a counter-intuitive possibility, e.g., a liquid that becomes unstable to gas-like fluctuations on cooling at positive pressure, has never been observed, neither in real substances nor in off-lattice simulations. More recently, such a re-entrant scenario has been dismissed on the premise that the re-entrant spinodal would collide with the gas-liquid coexisting curve (binodal) in the pressure-temperature plane. Here we study, numerically and analytically, two previously introduced one-component patchy particle models that both show (i) a re-entrant limit of stability of the liquid phase and (ii) a re-entrant binodal, providing a neat *in silico* (and *in charta*) realization of such unconventional thermodynamic scenario. *Published by AIP Publishing.* [<http://dx.doi.org/10.1063/1.4974830>]

Liquids with isobaric density extrema are particularly fascinating. The tendency to expand on cooling signals the onset of a structural ordering at microscopic level that overrides the ubiquitous densification originating from the reduced amplitude of thermal vibrations. The investigation of the thermodynamic behavior of liquids with density anomalies has clarified that density extrema are never isolated anomalies but are always associated with non-monotonic behaviors of several other response functions. In the case of water — the most common member of the group, whose temperature (T) of maximum density (TMD) at ambient pressure (P) is 4 °C — experimental studies have shown that, upon cooling, density, energy, and entropy fluctuations all increase, leading to an anomalous growth of the constant- P compressibility and specific heat.^{1–3} Such growth is further enhanced in supercooled states, where water is metastable with respect to crystallization.

The origin of density anomalies in water has animated the debate in the scientific community.^{1,2,4–18} Different thermodynamic consistent scenarios have been proposed^{19–22} leading to intense discussions which extend up to present days.^{23–28} The first thermodynamic scenario coherently accounting for the observed density, compressibility, and specific heat anomalies of water was proposed in 1982 by Robin Speedy.²⁹ In this very elegant piece of work, Speedy focused on the limit of stability of the liquid phase — which in mean-field coincides with the gas-liquid spinodal line — a line emanating from the gas-liquid critical point. In standard liquids, as predicted by the venerable van der Waals theory, the gas-liquid spinodal is

monotonic in the P – T plane approaching $T = 0$ at negative P , at the maximum tensile strength value.³ Speedy noticed that the spinodal line $P_{sp}(T)$ is an envelope of isochores^{3,30} and thus a TMD locus which intersects the spinodal line requires $dP_{sp}/dT = 0$ at the intersection, e.g., a re-entrant behavior. Fig. 1 shows a sketch of the Speedy phase diagram for positively and negatively sloped TMD loci. Albeit highly counterintuitive, if the spinodal traces back to positive P , the liquid becomes unstable to gas-like fluctuations both on heating and on cooling. In Speedy's scenario, the encounter of the retraced spinodal on cooling explained the observed increase in the response functions.

In 2003, Debenedetti³¹ called attention on the fact that a re-entrant spinodal cannot intersect with the metastable continuation of the liquid-gas coexisting curve without terminating there in an additional critical point. In the case of water, this would suggest that the response function should not display any significant increase for pressure sufficiently higher than the triple point pressure, at odd with experimental observation. The interest in finding realizations of the Speedy re-entrant spinodal has been progressively attenuated, if not suppressed, by the difficulty to imagine a liquid that would vaporize on cooling and by the availability of different thermodynamic scenarios equally able to rationalize the anomalies (the liquid-liquid critical point²¹ and the singularity-free²⁰ scenarios). In addition, simulations based on classical models of water do not show any retracing.^{30,32,33} Thus, with the exception of lattice models of water-like fluids,^{34–36} where the spinodal was found to retrace but only at negative pressures, Speedy's hypothesis remains a fascinating scenario that has never been realized, not even *in silico*.

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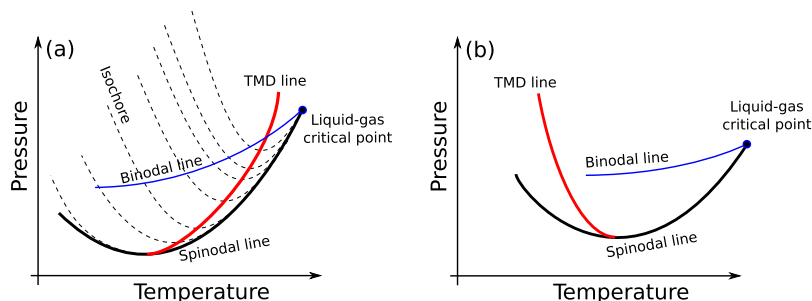


FIG. 1. (a) The re-entrant liquid-gas spinodal scenario. The TMD line connects all the minima of the isochores where $(\partial P/\partial T)_V = 0$ and intersects the spinodal line in its turning point. (b) The re-entrant liquid-gas spinodal scenario with negative sloped TMD line. The slope of the TMD line is related to the isobaric behavior of the isothermal compressibility $K_T = -(1/V)(\partial V/\partial P)_T$,²⁰ in turn proportional to the density fluctuations. If the TMD has a negative/positive slope in the T - P plane, then K_T increases upon cooling/heating.

In this communication we fill this gap, reporting two examples of a one-component system exhibiting re-entrant limit of stability of the liquid phase, covering both the cases presented in Fig. 1, e.g., with positively and negatively sloped TMD, respectively. Interestingly, in both cases, the intersection between the re-entrant metastability limit and the gas-liquid binodal is avoided, providing support to Speedy's scenario and its compatibility with Debenedetti's arguments.

The systems considered here are composed of spherical hard particles complemented with anisotropic attractions. In both cases, particles are modelled as hard spheres of diameter σ (the unit of length) and volume $v_s = \frac{\pi}{6}\sigma^3$. Each particle is decorated with n_p patches, which are modelled as truncated spherical cones and interact between themselves through a Kern-Frenkel (KF) potential,³⁷ an angular square well of depth ϵ (the unit of energy). See the [supplementary material](#) for a detailed presentation of the two models. Both models are examples of *patchy* particles,³⁸ colloids of new generation that have been shown to exhibit interesting and unique states such as empty liquids,³⁹ open crystals,^{40,41} water-like liquid-liquid critical points,⁴² and more.^{43–45}

Figure 2 shows a cartoon of the two models and the associated schematic phase diagram. Both models have been previously introduced and their phase diagram in the T - ρ plane has been evaluated. Here we return to these two models to calculate the equation of state and the TMD and stability loci in the P - T plane. The first model we consider is a Janus particle composed of two different hemispheres: one repulsive and one attractive. The hard sphere is decorated with a single patch covering half of its surface. The phase behaviour of Janus colloids is extremely rich, featuring non-crystalline ordered phases, cluster phases, crystalline lattices, and a gas-liquid phase transition.^{46–49} The latter has a peculiar shape as the two coexisting densities, gas and liquid alike, increase

upon cooling. However, the coexisting gas density increases faster, leading to the shrinkage, and possibly closure, of the unstable region. However, the seemingly inevitable appearance of a lower critical point is prevented by the presence of an ordered lamellar phase (see Fig. 2).⁴⁷ In the second model,^{50,51} indicated as 2A9B in the following, the hard sphere is decorated with two patches of type A, located on opposite hemispheres in such a way that the patch-particle center-patch angle is γ , and nine patches of type B, equi-spaced on the equator. Bonds between B patches are disabled. The attraction strengths of the interactions are chosen in such a way that the formation of AA bonds, and hence of long chain-like structures, is energetically favoured. However, the larger number of B-patches makes AB bonds entropically convenient, promoting the branching of the chains. When $\gamma = 180^\circ$, the competition between the two mechanisms results in a pinched gas-liquid phase diagram, where the density of the liquid phase decreases as the temperature goes down.⁵⁰ For smaller values of γ , however, chains are much more flexible and the system tends to form rings at very low T . These rings stabilise the gas phase, causing a re-entrance of the density of the gas which approaches the coexisting liquid one. The net effect is to generate a closed gas-liquid coexistence loop in the T - ρ plane with two critical points (see Fig. 2).⁵¹ Here we choose $\gamma = 130^\circ$. The Janus and 2A9B models investigated here display gas-liquid phase separation regions where, at low T , the ρ of one or both of the two phases does not tend to a constant value, but rather continuously increases or decreases as shown in Fig. 2. As a result, the two phases approach each other, and the phase diagram displays a *re-entrant* shape.⁵² As we will show in the following, such a phenomenon, which is linked to the peculiar self-assembly processes occurring in these systems, is also connected to thermodynamic anomalies.

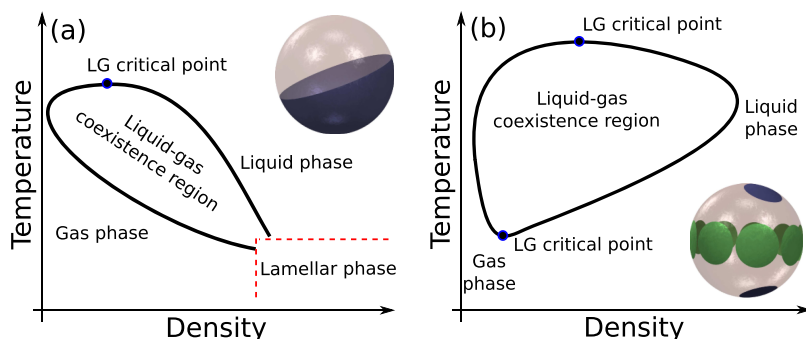


FIG. 2. (a) Sketch of the phase diagram in the ρ - T plane for the Janus system, composed of particles that are hard spheres with an attractive patch covering half of its surface (in blue). (b) Sketch of the phase diagram of the 2A9B system. The surface of these particles is decorated with two patches of type A (in blue) located close to the poles, so that the relative patch-center-patch angle is 130° , and nine patches of type B (in green) equispaced on the equator.

To evaluate $P(\rho, T)$, we perform simulations in the grand canonical ensemble with the successive umbrella sampling (SUS) technique,⁵³ which consists in an independent sampling of overlapping density intervals. The results obtained in each interval at fixed T , volume V , and chemical potential μ are then stitched together by using the overlapping regions between neighbouring windows.⁵³ The main simulation output is the probability that the system has a number density $\rho = N/V$ (with N the number of particles in the system), namely, $\mathcal{N}(\rho; z)$, computed at fixed V , T , and activity $z = e^{\beta\mu}$, where $\beta = 1/k_B T$ and k_B is the Boltzmann constant. The uniform (in N) sampling provided by the SUS technique allows us to evaluate $\mathcal{N}(\rho; z)$ at different μ by employing histogram reweighting techniques.⁵⁴

As an example, Figure 3(a) shows the $\mathcal{N}(\rho; z)$ curves obtained for the Janus system. All the curves have been reweighted at coexistence, which is reached when the area below each of the two peaks is the same. From $\mathcal{N}(\rho; z)$, $P(\rho, T)$ can be directly calculated (see the [supplementary](#)

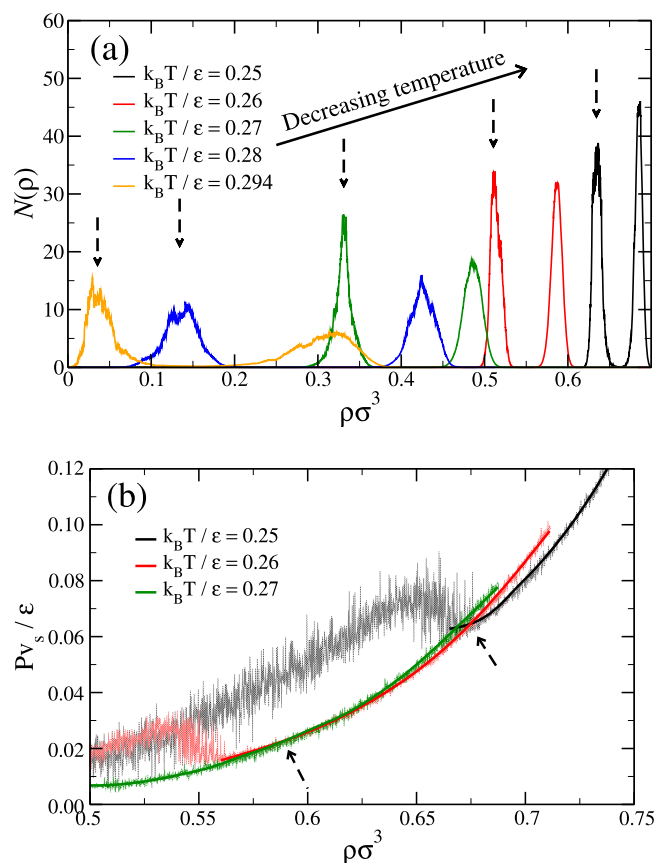


FIG. 3. (a) Probability distributions of the density, $\mathcal{N}(\rho; z)$, for the Janus system at five different T . The bimodality of the curves demonstrates that, at all these T , there is a phase separation between a low-density (gas) and a high-density (liquid) phase. The vertical dashed arrows indicate the position of the peak relative to the gas phase, which moves to higher and higher density as T decreases. The value of the chemical potential at each T has been chosen so that the area below each curve is equally shared between the two peaks, which is the condition of phase coexistence. The simulation box has a side $L = 20\sigma$, requiring the investigation of a number of particles extending from 0 up to 5000. (b) Equations of state $P(\rho, T)$ for the Janus system at three different T . The raw data (shaded curves) are splined under tension. The intersections between the resulting curves (solid lines), here indicated with arrows, yield estimates for the location of the density maxima.

material). As shown by Binder and his group,⁵⁵ when simulations are properly performed and the system has been able to equilibrate even inside the coexistence region, $P(\rho, T)$ provides a consistent thermodynamic description valid for the investigated system size. From the resulting $P(\rho, T)$, we thus estimate the coexistence pressure as well as the lowest pressure on the liquid branch, which is associated with a bubble evaporation/condensation transition and it is used here as a limit of stability of the liquid phase and as a proxy of the mean-field spinodal.^{55–57} Finally, from the intersection of the curves $P(\rho, T)$ for different values of T we determine the presence of isobaric extrema of the density ρ (Fig. 3(b) and the [supplementary material](#)).

Figure 4 shows the binodal, the limit of stability, and TMD lines, as evaluated for the Janus model in the P – T plane. We first note that, at high T , the limit of stability and binodal lines originate from the critical point. As T decreases, the binodal P remains roughly constant, while P_{sp} decreases, becoming negative. At around $k_B T / \epsilon \approx 0.2775$ a minimum in P_{sp} appears, and then both the limit of stability and binodal P quickly raise up as the system is further cooled down. By comparing these results with the T – ρ phase diagram,⁴⁷ we link the observed sudden increase of the coexistence pressure on cooling to the increasing of the coexisting density of the gas. Indeed, in this T region the coexisting gas, which at higher T is a diluted phase made of mostly non-interacting particles, turns into a cluster phase whose basic constituents, micelles and vesicles, are finite-sized aggregates of particles.⁴⁷ The return of the limit of stability line to the $P > 0$ semi-plane, which happens around $k_B T / \epsilon \approx 0.274$, is the first numerical example of a 3D off-lattice system exhibiting a Speedy-like scenario.

Our results show that the Janus model also exhibits a line of density maxima, as required by thermodynamic consistency.³⁰ This TMD line, shown in Figure 4, has a negative slope, as in water.⁵⁸ It starts from high pressure at low temperature and quickly goes down, intercepting the limit of stability line. The two lines meet, within our numerical accuracy, exactly at the minimum of the limit of stability curve, as highlighted in the inset of Figure 4. At the intersection point, the

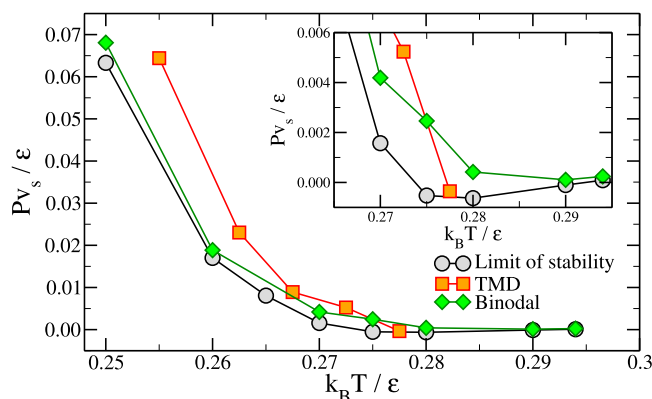


FIG. 4. The limit of stability (gray circles), TMD (orange squares), and binodal (green diamonds) loci of the Janus model in the P – T plane. (Inset) A blow-up of the main plot, showing that the TMD line ends in the minimum of the limit of stability line. The error bars associated with the limit-of-stability pressure are smaller than the symbols' sizes. Lines are guides for the eye.

TMD line shows a vanishing slope, consistently with thermodynamic predictions.³⁰ It is interesting to note that the sheer existence of a binodal with a negative slope shows that it is possible for a re-entrant spinodal to not intersect the binodal line.³¹

Next we focus on the 2A9B model. To favour the formation of energetically stable weakly interacting aggregates (rings as opposed to chains, e.g., structures in which all AA bonds are satisfied) in the gas-phase, we select $\gamma = 130^\circ$. This provides thermodynamic stability to the gas phase.⁵¹ Figure 5 shows the thermodynamics *loci* of the model in the P - T plane. In contrast to the Janus model, the pressure of the binodal (for $k_B T/\epsilon > 0.045$) and of the TMD is an increasing (rather than decreasing) function of T . Similarly to the Janus case, the limit of stability of the liquid phase is also non-monotonic, with a minimum occurring at $k_B T/\epsilon \approx 0.055$. As a result, the TMD and the limit of stability curves meet, within our numerical accuracy, in the minimum of the latter, consistently with thermodynamic predictions. In contrast with the Janus case, however, the TMD line has a positive slope.

The 2A9B model with $\gamma = 130^\circ$ can also be solved in mean-field,⁵¹ within the Wertheim formalism,⁵⁹ augmented with the inclusion of closed ring loops.^{51,60} The analytic solution provides an unambiguous definition of the mean-field spinodal line and thus implicitly a check of the numerical definition. The theoretical results, shown as full lines in the figure, compare well with simulation data in the T -range where this is available. At lower T the theory shows that the spinodal curve returns to positive P , fully consistent with Speedy's scenario. According to the theory (and with Debenedetti's arguments), the spinodal line ends at the lower critical point ($k_B T/\epsilon \approx 0.0339$), where it meets with the binodal.

To summarise, a retracing LG spinodal was proposed for the first time almost 35 years ago by Speedy to explain the low-temperature anomalies of water.¹⁹ In this communication, for the first time, we have presented two off-lattice examples of systems exhibiting such a unique thermodynamic feature. In both cases, the liquid vaporizes on cooling, but the gas phase

is composed by (weakly interacting) aggregates in which particles are ordered in configurations of very low energy but also very low entropy, changing the slope of the binodal at low T . In the Janus case, the gas is indeed formed by micelles and vesicles,⁴⁷ while in the 2A9B case, the gas is composed of rings.⁵¹ Thus, the physics that stabilizes the gas phase at low T allows the spinodal to rise back to positive P . Finally, we note that both models might be realized experimentally in the near future,^{61–64} allowing for an experimental confirmation of the numerical and theoretical results reported here. The peculiar properties of these systems, while being somewhat different from those of water (for which there is no evidence of a lower gas-liquid critical point and the binodal is positively sloped), can shed light on the thermodynamics of anomalous fluids. We have found that the observed non-monotonic behaviour of the LG spinodal line is linked to the re-entrance of the density of the coexisting gas (Fig. 2). This strongly suggests that such re-entrance is a sufficient condition to observe a non-monotonic spinodal line and, as a consequence of thermodynamic consistency, a TMD line which meets the spinodal exactly in its minimum.

See the [supplementary material](#) for additional details on the models and methods and for simulation snapshots of the gas-stabilising aggregates.

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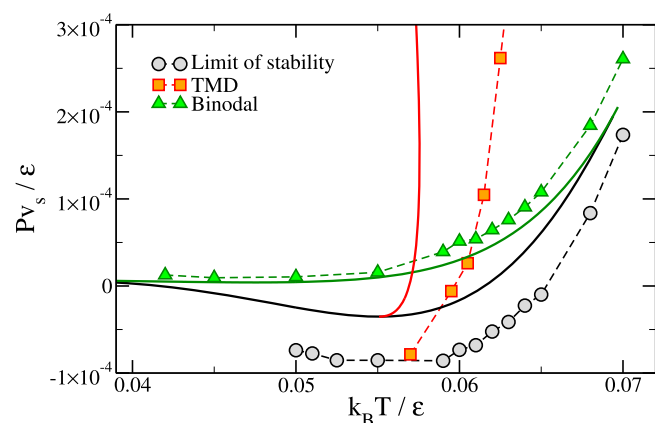


FIG. 5. Theoretical and numerical results for the 2A9B model: limit of stability (black), TMD (orange), and binodal (green) loci in the P - T plane. Symbols indicate numerical results, while theoretical curves are shown as full lines. The error bars associated with the limit-of-stability pressure are smaller than the symbols' sizes. Dashed lines are guides for the eye. Numerical results refer to a system with $L = 14\sigma$.

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