

PAPER • OPEN ACCESS

## Elastic Differential Cross Sections for Electron Scattering with Dichloromethane

To cite this article: E. Lange *et al* 2017 *J. Phys.: Conf. Ser.* **875** 062036

View the [article online](#) for updates and enhancements.

### Related content

- [FTIR and QCL diagnostics of the decomposition of volatile organic compounds in an atmospheric pressure dielectric packed bed plasma reactor](#)  
Z Abd Allah, D Sawtell, V L Kasyutich et al.
- [Comparison of Conventional and Microwave-assisted Synthesis of Benzimidazole Derivative from Citronella in Kaffir lime oil \(Citrus hystrix DC.\)](#)  
W Warsito, A.S Noorhamdani, Suratmo et al.
- [Lauryl Amine as heavy metal collector of boiler ash from pulp and paper mill waste](#)  
M P Sembiring, J Kaban, N Bangun et al.

## Elastic Differential Cross Sections for Electron Scattering with Dichloromethane

E. Lange\*, K. Krupa<sup>‡</sup>, J. Ameixa\*, A. S. Barbosa<sup>†</sup>, D. F. Pastega<sup>†</sup>, P. Limão-Vieira\*, M. H. F. Bettega<sup>†</sup>, F. Blanco<sup>§</sup>, G. García<sup>‡</sup>, F. Ferreira da Silva<sup>\*†</sup>

\* Laboratório de Colisões Atômicas e Moleculares, CEFITEC, Departamento de Física, Faculdade de Ciências e Tecnologia, Universidade NOVA de Lisboa, 2829-516 Caparica, Portugal

<sup>†</sup> Departamento de Física, Universidade Federal do Paraná, 81531-990, Curitiba, Paraná

<sup>§</sup> Departamento de Física Atómica, Molecular y Nuclear, Universidad Complutense de Madrid, Avenida Complutense, 28040 Madrid, Spain

<sup>‡</sup> Instituto de Física Fundamental, Consejo Superior de Investigaciones Científicas, Serrano 113-bis, 28006 Madrid, Spain

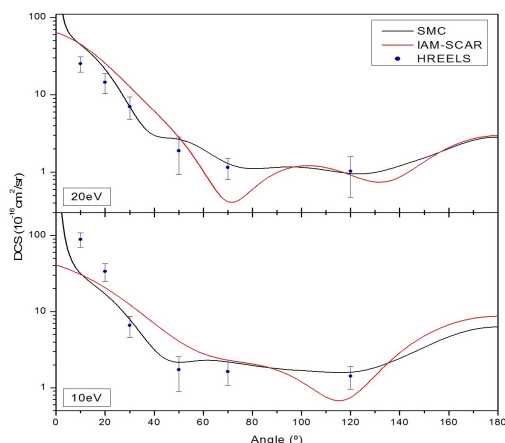
**Synopsis** In the present study joint experimental and theoretical elastic differential cross sections for electron scattering from dichloromethane in the incident electron energy region 7 to 50 eV are discussed.

Dichloromethane ( $\text{CH}_2\text{Cl}_2$ ) is a relevant atmospheric and environmental molecule, where its high volatility results from the constant use in chemical industries [1], in biomass production [2] as well as from oceanic emissions [3]. Once in the stratosphere, the main sink mechanism has been attributed to photolysis leading to chlorine radical formation. Such radical at tropospheric altitudes further acts as a catalyst to ozone fragmentation into  $\text{ClO}$ .

Photolysis [4], photoabsorption [5] and DEA [6] studies can be found in the literature, but only a few related with electron scattering processes and even those are in the high energy regime [7] or strictly theoretical approaches [8, 9].

were obtained in a High Resolution Electron Energy Loss Spectrometer (HREELS) [10] with an energy resolution of 120 meV (FWHM). The theoretical calculations were performed with two different methodologies: the Schwinger Multichannel Method (SMC) implemented with pseudopotentials [11] and the Independent Atom Method with Screening Corrected Additivity Rule (IAM-SCAR)[12]. The SMC method presents a better description of the elastic scattering at lower electron impact energies (up to 20 eV), where the Born Closure correction for long range potentials gives a good description of the dipolar cross section dependence for smaller scattering angles. The IAM-SCAR method shows a better description for electron impact energies above 20 eV.

The excellent agreement between experimental DCSs and both theoretical approaches leads to a good description of the shapes and angular distribution of the cross sections.



**Figure 1.** Elastic DCS for electron scattering from  $\text{CH}_2\text{Cl}_2$  at 10 and 20 eV electron impact energy.

Here we present a comprehensive joint experimental and theoretical study on differential cross sections (DCSs) for elastic electron scattering from  $\text{CH}_2\text{Cl}_2$  molecule for incident electron energies 7, 10, 20, 30 and 50 eV. The experimental DCSs

## References

- [1] K. F. Alcantara *et al.* 2011 *J. Phys. B* **44** 165205
- [2] J. M. Lombert *et al.* 1999 *J. Geophys. Res.* **104** 8373
- [3] M. A. K. Khalil in *Reactive halogen compounds in the atmosphere* Springer; 1999. **4E** p. 45-79
- [4] M. Yen *et al.* 1993 *J. Chem. Phys.* **99** 123
- [5] A. Mandal *et al.* 2014 *J. Quant. Spec. Rad.* **149** 291
- [6] G. A. Gallup *et al.* 2011 *J. Chem. Phys.* **135** 134316
- [7] C. G. P. Karwasz *et al.* 1999 *Phys. Rev. A* **59** 1341
- [8] A. P. P. Natalense *et al.* 1999 *Phys. Rev. A* **59** 879
- [9] R. Naghma *et al.* 2014 *J. Elec. Spec. Rel. Phen.* **193** 48
- [10] F. MotteTollet *et al.* 1992 *J. Chem. Phys.* **97** 7314
- [11] R. F. da Costa *et al.* 2015 *Eur. Phys. J. D* **69** 159
- [12] F. Blanco *et al.* 2013 *Eur. Phys. J. D* **67** 199

<sup>†</sup>E-mail: f.ferreiradasilva@fct.unl.pt